

## Mastering Chemistry Answers Chapter 8

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### Mastering Chemistry Answers Chapter 8

Measurement of energy in a thermodynamic system. It is equal to the internal energy of the system plus the product of pressure and volume. Amount of thermal heat that flows when a reaction occurs at constant pressure. The direction of which thermal energy flows when the reaction occurs.

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Given that  $4 \text{ NH}_3 + 5 \text{ O}_2 \rightarrow 4 \text{ NO} + 6 \text{ H}_2\text{O}$ , if 3.00 mol  $\text{NH}_3$  were made to react with excess of oxygen gas, the amount of  $\text{H}_2\text{O}$  formed would be: A 24.0 g sample of nitrogen gas reacts with an excess of hydrogen gas to give an actual yield of 3.85 g  $\text{NH}_3$ .

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(d) Argon (Ar), a group 18 noble gas has 18 total electrons. The first two are in the orbital. The next two are in the orbital. Six are assigned to the orbital. Two more go into the orbital. The final six are in the orbital. Noble gasses have all of their orbitals filled with electrons. The total configuration of argon is .

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answer options - 1)) to discourage students from guessing. All other questions were typically a deduction of 8 percent per incorrect answer. Prior to lectures, Lufaso reviewed the diagnostics from the completed prelecture assignments in order to better understand what concepts students struggled with and to focus on those during class time.

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